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Students can watch the intro video and away they go. Switch solutes to compare different chemicals and find out how concentrated you can go before you hit saturation!PhET: Density How do strong and weak acids differ? PhET: Acid-Base Solutions How do strong and weak acids differ? Now includes both printable PDF and Google Slides versions!The PhET: Isotopes and Atomic Mass Activity Guide is used along with the free *Next Generation PhET Simulation "Isotopes and Atomic Mass". Students learn how to construct Isotope models for basic elements by adding neutrons one at a time and observing how this effects Mass Number, Atomic Mass, and stability of the Isotope created. Can a weak acid solution have the same pH as a strong acid solution?PhET: Atomic Interactions Explore the interactions between various combinations of two atoms. Create your own sandwich and then see how many sandwiches you can make with different amounts of ingredients. Find out by adding single, double or triple bonds and lone pairs to the central atom. Then see how concentration and strength affect pH. Investigate how adding more of a liquid or diluting with water affects pH.PhET: Reactants, Products and Leftovers Create your own sandwich and then see how many sandwiches you can make with different amounts of ingredients. View the color of the peak of the spectral curve.PhET: Build an Atom Build an atom out of protons, neutrons, and electrons, and see how the element, charge, and mass change. The best part is that they don't even know they are learning, they just think they are having fun!The Isotopes and Atomic Mass Activity Guide will direct students through the simulation leading them to experiment with changing the number of neutrons and discovering how this forms various Isotopes of an element. Use lab tools on your computer to find out! Dip the paper or the probe into solution to measure the pH, or put in the electrodes to measure the conductivity. Observe the the total force acting on the atoms or the individual attractive and repulsive forces. Simulate the famous experiment in which he disproved the Plum Pudding model of the atom by observing alpha particles bouncing off atoms and determining that they must have a small core.PhET: States of Matter Explore the interactions between various combinations of two atoms. Play a game to test your understanding of reactants, products and leftovers. This lesson is perfect for face to face, hybrid, or remote learning! This product includes the Activity Guide (pdf), Activity Guide (Google Slides), Answer Key (pdf), and a Getting Started Guide for teachers.*PhET simulations are fun, interactive, research-based simulations of physical phenomena. Students also learn the relationship between Relative Abundance of Isotopes in Nature and the Average Atomic Mass of Elements on the Periodic Table. Play a game to test your ideas!PhET: Beer's Law Lab "The thicker the glass, the darker the brew, the less the light that passes through." Make colorful concentrated and dilute solutions and explore how much light they absorb and transmit using a virtual spectrophotometer!PhET: Blackbody Spectrum How does the blackbody spectrum of the sun compare to visible light? What affects the rate of a reaction?PhET: Rutherford Scattering How did Rutherford figure out the structure of the atom without being able to see it? The new Next Generation sims utilize HTML 5 which allows them to run in any modern web browser on most devices including Chromebooks and tablets like the iPad! Learn more about using PhET Simulations with your students.Bundle and save!PhET Activity Guides Chemistry BUNDLE More PhET simulation Activity Guides:PhET: Build an Atom Activity GuidePhET: Reactants, Products, and Leftovers Activity GuidePhET: Build a Molecule Activity GuidePhET: Collision Lab Activity GuidePhET: Projectile Motion Activity GuidePhET: Eating and Exercise Activity GuidePhET: Lunar Lander Activity GuidePhET: Balancing Act Activity GuidePhET: Isotopes and Atomic Mass Activity Guide Staff Web Page Directory " 1-SchoolwiresTraining Albani, Andrew Balantic, Jeannette Bauer, Susan Beckner, Dolores Berg, Michael Bilello, Steven Blum, Loren Bombardiere, Dina Brunetti, Lea Burdi, Cathy Calemmo, Margaux Caputo, Megan Cardella, Christina Cardilicchia, Sandra Casale, Tim Cascio, Elena Cerar, Zoran Cerrone, Dawn Chaputian, Ray Cloghessy, Anne Cohen, Sue Colwell, Jeanne Hauser, Amanda Cortese, Jean Cotter, James Crawford, Jennifer Cutter, Paul DeColibus, Michele DellaMonica, Daniel Demorest, Barbara Department, Library Department, Music Department, Nurse DiBitetto, Kevin Dolan, Diane Doyle, Patrick Dubuke, Kathleen Dunleavy, James Dunne, Lauren Durkin, Kathleen Ellis, Scott Esposito, Domenick Ettinger, David Farkash, Stephanie Fediw, Dennis Ferraiolo, Crystal Ferrara, Mary Finnell, Stephen Flaherty, Tori Fleming, Nancy Foley, Meredith Gallagher, Nancy Galvez, Andrea Gerbasi, Philip Getchell, Donna Giacalone, Peter Giannotta, Kathleen Going, Marilyn Gordon, Steven Grisafi, Marie Gutierrez, Antonietta Harrison, George Heedles, Michael Hepworth, Brad Highfill, William Holz, Denisse Imbesi, Maureen Infranca, James Intintoli, Sebastian Izzo, David Jonas, Beth Karp, Jason Kear, Tammy Kemnitzer, Michele Kennedy, Jeanne Knopf, Janine Korn, Michael Landrigan, Jennifer Lasher, Kathryn Maglio, Darlene Malaszczysz, Lisa Malone, James Mann, Lavleen Manning, Cynthia Mannion, Patricia Marino, Amanda Masry, Rivka Mayo, Stephen Mazzella, Joan McAuley, Scott McCrann, James McCrann, Robert McDonald, Brian McDonald, Christina McKinstry, Erin McLaughlin, Angela Mein, Alice Mendelson, Melina Menon, Beena Mich, Erin Modiano, Jeffrey Morris, Nicole Morris, Cindy Moscato, Annmarie Moss, Allison Nakutavicius, Tara Napoli, Christopher Nocella, Thomas O'Hagan, Kevin Office, Main Olsen, Randall Oriani, Catherine Pane, Vincent Perez, Luis Perlow, Heather Perrotta, Dave Pierson, Amanda Poisson, Anne Prasso, Nina Quigley, Mark Rebolini, Carlo Regis, Amy Reichman, Marisa Rigley, Kathryn Rogler, Janet Rupes, Susanne Russo, Carol Santoro, Peter Scala, Mary Ellen Scanlon, Anne Sckalor, Kathryn Sclafani, Reid Scoria, John Seeley, Kristina Sergi, Trish Sheil, Kathryn Skolnik, Samara Slater, John Smith, Joan Squires, Katherine Stano, Michael Stratis, Michael Strunk, Kathleen Tanzosh, Cynthia Tavernise, Suzanne Tierney, Marilyn Tomaini, Steven Uttendorfer, Shawn Vicchiarelli, Michael Wallace, Alison Walley, Scott Wiedenhoff, Robert Wolf, Darryl Young, Diana Zagari, Michael The PhET Interactive Simulations project at the University of Colorado Boulder creates free interactive math and science simulations. Change solutes to compare different chemical compounds in water.PhET: Molecules and Light Do you ever wonder how a greenhouse gas affects the climate, or why the ozone layer is important. How can you tell one isotope from another? Just print the activity guide or share from your Google Drive. Then, compare the model to real molecules!PhET: Molecule Shapes: Basics Explore molecule shapes by building molecules in 3D! Find out how a molecule's shape changes as you add atoms to a molecule.PhET: pH Scale Test the pH of things like coffee, spit, and soap to determine whether each is acidic, basic, or neutral. Use the sim to explore how light interacts with molecules in our atmosphere.PhET: Molecule Polarity PhET: Molecule Shapes Explore molecule shapes by building molecules in 3D! How does molecule shape change with different numbers of bonds and electron pairs? Switch between logarithmic and linear scales. See how many PhET: Reactions & Rates Explore what makes a reaction happen by colliding atoms and molecules. Measure the temperature and pressure, and discover how the properties of the gas vary in relation to each other.PhET: Isotopes and Atomic Mass Are all atoms of an element the same? What can you change to balance an equation? When are reactions reversible? Customize the attraction to see how changing the atomic diameter and interaction strength affects the interaction.PhET: Balancing Chemical Equations How do you know if a chemical equation is balanced? Then check molarity with the concentration meter. As students interact with the simulation, they get immediate feedback about the affects of the changes they made. Students will be introduced to the following concepts:isotopesatomic massaverage atomic massrelative stable and unstable isotopescalculating average atomic massThe activity guide begins with a short introductory video to give students background information and includes both PDF and Google Slides versions for printing worksheets or sharing via Google Drive or Google Classroom. Your students will hit the ground running without the need for a lengthy introduction from you. Use the sim to learn about isotopes and how abundance relates to the average atomic mass of an element.PhET: Molarity What determines the concentration of a solution? Or you can design your own liquid!PhET: pH Scale: Basics Test the pH of everyday liquids such as coffee, spit, and soap to determine whether each is acidic, basic, or neutral. Adjust the temperature to see the wavelength and intensity of the spectrum change. Investigate whether changing the volume or diluting with water affects the pH. All PhET simulations are freely available from the PhET website (phet.colorado.edu) and are easy to use and incorporate into the classroom. See how many products you can make with different amounts of reactants. Do the same with chemical reactions. Customize the attraction to see how changing the atomic diameter and interaction strength affects the interaction.PhET: Under Pressure Explore pressure under and above water. Visualize the relative number of hydroxide ions and hydronium ions in solution. Can a weak acid solution have the same pH as a strong acid solution?PhET: Gas Properties Pump gas molecules to a box and see what happens as you change the volume, add or remove heat, change gravity, and more. Then play a game to test your ideas!PhET: Concentration Watch your solution change color as you mix chemicals with water. PhET sims are based on extensive education research and engage students through an intuitive, game-like environment where students learn through exploration and discovery. Can you get a perfect score on each level? What are all the ways you can change the concentration of your solution? Learn about the blackbody spectrum of the sun, a light bulb, an oven, and the earth. Learn about the relationships between moles, liters, and molarity by adjusting the amount of solute and solution volume. Design experiments with different reactions, concentrations, and temperatures. See how pressure changes as you change fluids, gravity, container shapes, and volume.

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